

LASER-PRODUCED PLASMA SPECTROSCOPY OF MEDIUM TO HIGH-Z ELEMENTS IN THE 2 TO 9 NM SPECTRAL REGION

**Elaine Long¹, Chihiro Suzuki², John Sheil¹, Elgiva White¹,
Takamitsu Otsuka³, Ragava Lokasani¹, Bowen Li¹, Robert
Stefaniuk¹, Paul Sheridan¹, Paddy Hayden⁴, Emma Sokell¹, Gerry
O'Sullivan¹, Fergal O'Reilly¹ & Padraig Dunne¹**

¹Atomic, Molecular, Plasma Physics and Spectroscopy Group, School of Physics, University
College Dublin

²National Institute for Fusion Science, 322-6 Oroshi-cho, Toki 509-5292, Japan

³Utsunomiya University, Japan

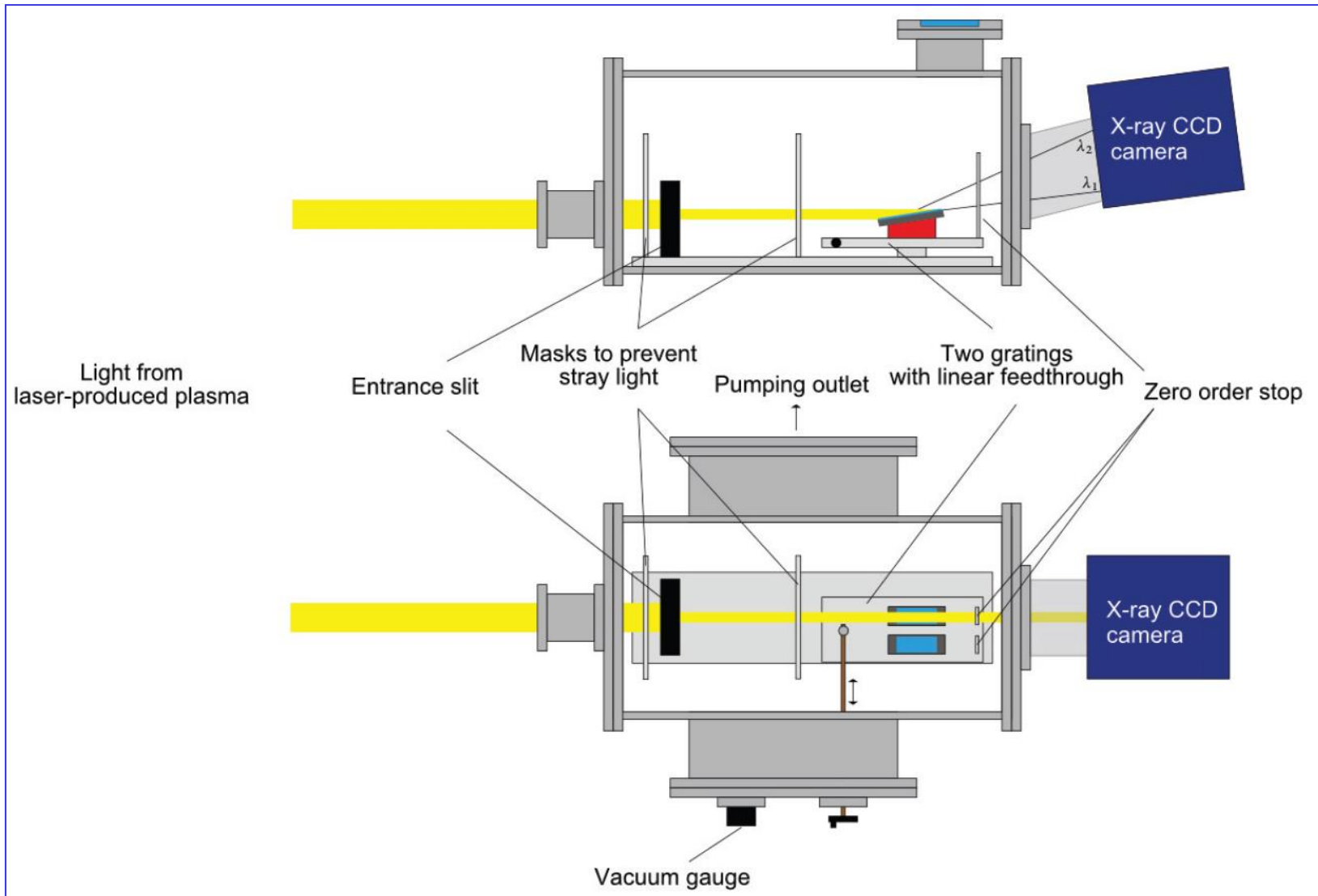
⁴National Centre for Plasma Science & Technology and School of Physical Sciences, Dublin City
University



Outline

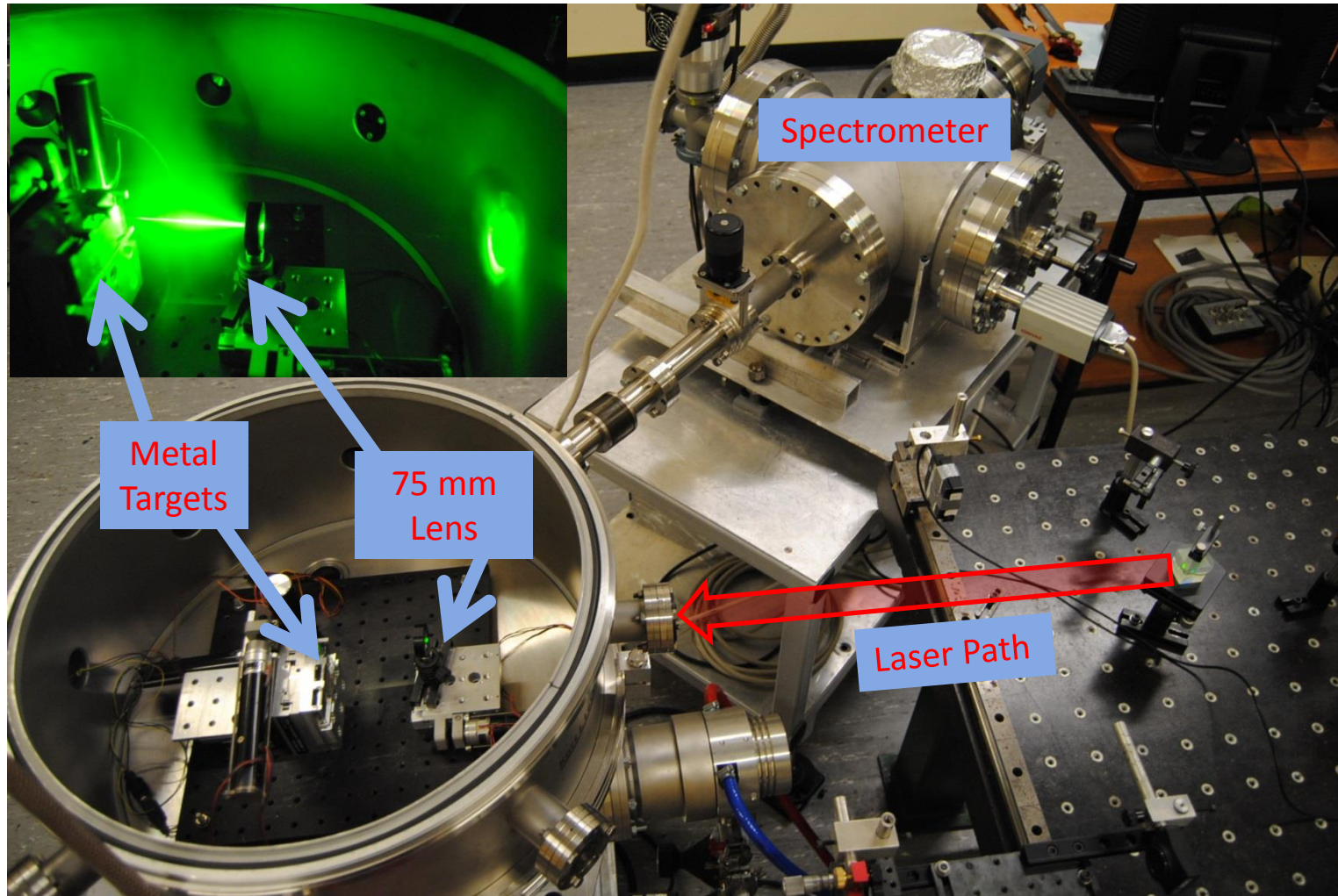
- Emission Spectroscopy
 - Experimental Set-up
 - Targets used
 - Spectra obtained
 - Cowan Code Calculations
- Absorption Spectroscopy
- Plasma Imaging

Schematic Diagram of Spectrometer



(courtesy of Takamitsu Otsuka)

Emission Spectroscopy Set-Up



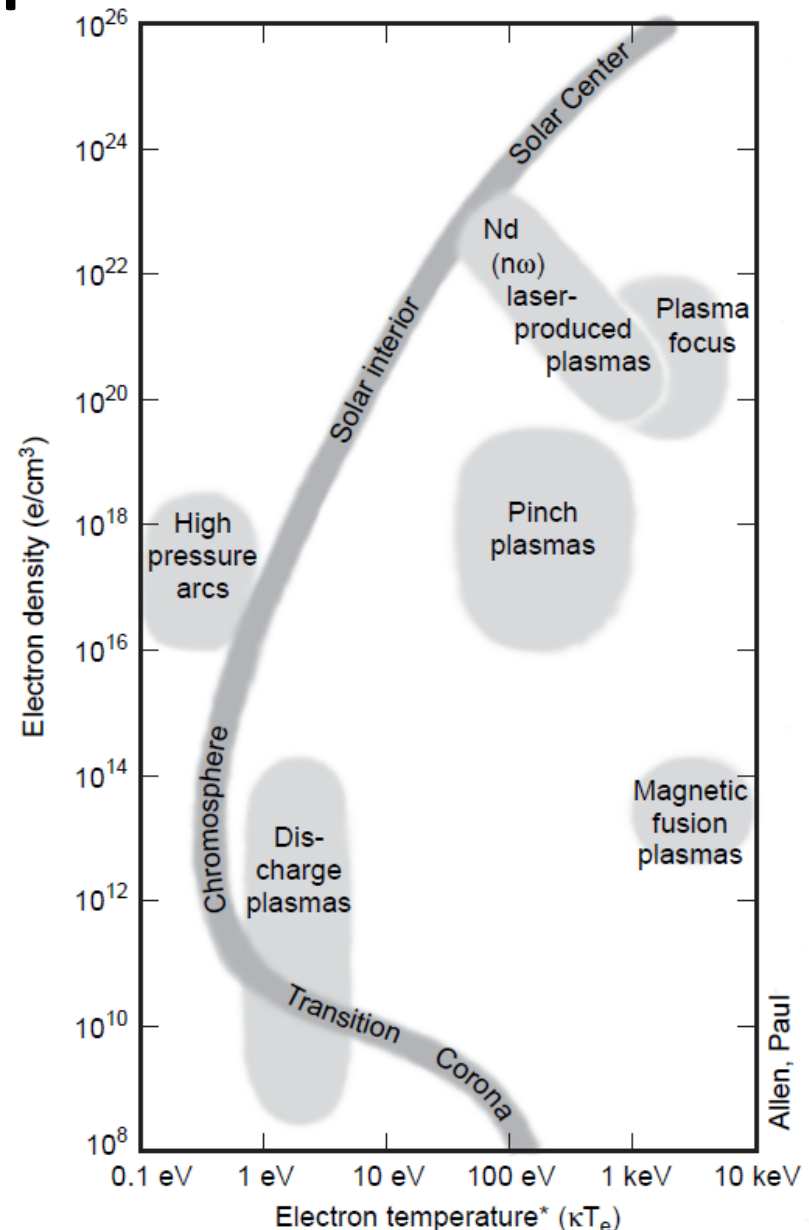
Plasma Properties

Temperature depends on laser power density (Φ)

$$T_e(\text{eV}) \approx (\lambda^2 \Phi)^{3/5}$$

Electron density $10^{19} - 10^{21}$ cm^{-3} depending on laser wavelength

$$n_{ec} \sim \frac{10^{21}}{\lambda^2} (\text{cm}^{-3})$$



Targets used for Emission Spectroscopy

spectroscopy

hydrogen 1 H 1.0079																	helium 2 He 4.0026
lithium 3 Li 6.941	beryllium 4 Be 9.0122											boron 5 B 10.811	carbon 6 C 12.011	nitrogen 7 N 14.007	oxygen 8 O 15.999	fluorine 9 F 18.998	neon 10 Ne 20.180
sodium 11 Na 22.990	magnesium 12 Mg 24.305											aluminium 13 Al 26.982	silicon 14 Si 28.086	phosphorus 15 P 30.974	sulfur 16 S 32.065	chlorine 17 Cl 35.453	argon 18 Ar 39.948
potassium 19 K 39.098	calcium 20 Ca 40.078	scandium 21 Sc 44.956	titanium 22 Ti 47.867	vanadium 23 V 50.942	chromium 24 Cr 51.996	manganese 25 Mn 54.938	iron 26 Fe 55.845	cobalt 27 Co 58.933	nickel 28 Ni 58.693	copper 29 Cu 63.546	zinc 30 Zn 65.39	gallium 31 Ga 69.723	germanium 32 Ge 72.61	arsenic 33 As 74.922	selenium 34 Se 78.96	bromine 35 Br 79.904	krypton 36 Kr 83.80
rubidium 37 Rb 85.468	strontium 38 Sr 87.62	yttrium 39 Y 88.906	zirconium 40 Zr 91.224	niobium 41 Nb 92.906	molybdenum 42 Mo 95.94	technetium 43 Tc [98]	ruthenium 44 Ru 101.07	rhodium 45 Rh 102.91	palladium 46 Pd 106.42	silver 47 Ag 107.87	cadmium 48 Cd 112.41	indium 49 In 114.82	tin 50 Sn 118.71	antimony 51 Sb 121.76	tellurium 52 Te 127.60	iodine 53 I 126.90	xenon 54 Xe 131.29
caesium 55 Cs 132.91	barium 56 Ba 137.33	57-70 ✱	lutetium 71 Lu 174.97	hafnium 72 Hf 178.49	tantalum 73 Ta 180.95	tungsten 74 W 183.84	rhenium 75 Re 186.21	osmium 76 Os 190.23	iridium 77 Ir 192.22	gold 79 Au 196.97	mercury 80 Hg 200.59	thallium 81 Tl 204.38	lead 82 Pb 207.2	bismuth 83 Bi 208.98	polonium 84 Po [209]	astatine 85 At [210]	radon 86 Rn [222]
francium 87 Fr [223]	radium 88 Ra [226]	89-102 ✱ ✱	lawrencium 103 Lr [262]	rutherfordium 104 Rf [261]	dubnium 105 Db [262]	seaborgium 106 Sg [266]	bohrium 107 Bh [264]	hassium 108 Hs [269]	meitnerium 109 Mt [268]	ununnillium 110 Uun [271]	unununium 111 Uuu [272]	ununbium 112 Uub [277]	ununquadium 114 Uuq [289]				

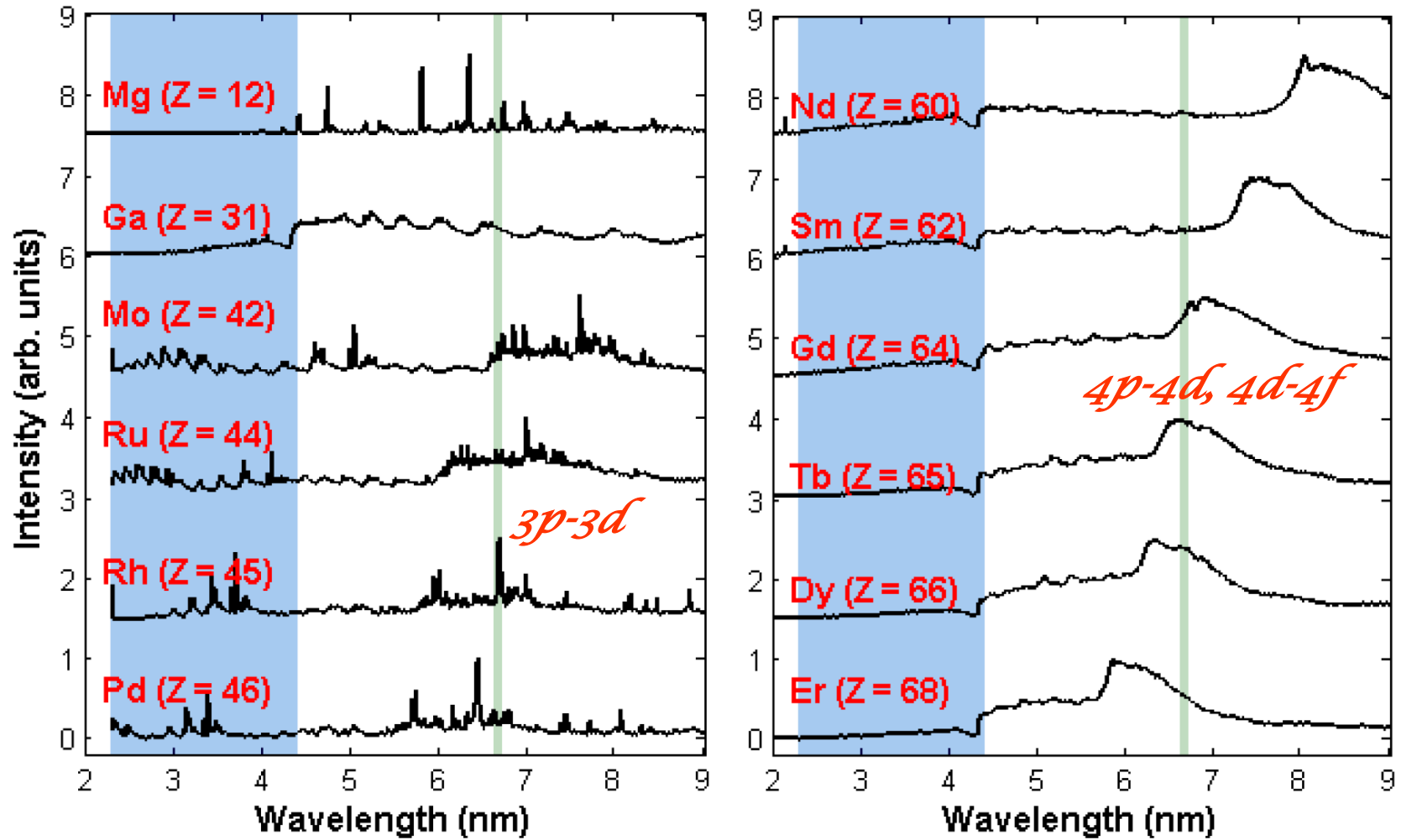
* Lanthanide series

** Actinide series

lanthanum 57 La 138.91	cerium 58 Ce 140.12	praseodymium 59 Pr 140.91	neodymium 60 Nd 144.24	promethium 61 Pm [145]	samarium 62 Sm 150.36	europium 63 Eu 151.96	gadolinium 64 Gd 157.25	terbium 65 Tb 158.93	dysprosium 66 Dy 162.50	holmium 67 Ho 164.93	erbium 68 Er 167.26	thulium 69 Tm 168.93	ytterbium 70 Yb 173.04
actinium 89 Ac [227]	thorium 90 Th 232.04	protactinium 91 Pa 231.04	uranium 92 U 238.03	neptunium 93 Np [237]	plutonium 94 Pu [244]	americium 95 Am [243]	curium 96 Cm [247]	berkelium 97 Bk [247]	californium 98 Cf [251]	einsteinium 99 Es [252]	fermium 100 Fm [257]	mendelevium 101 Md [258]	nobelium 102 No [259]

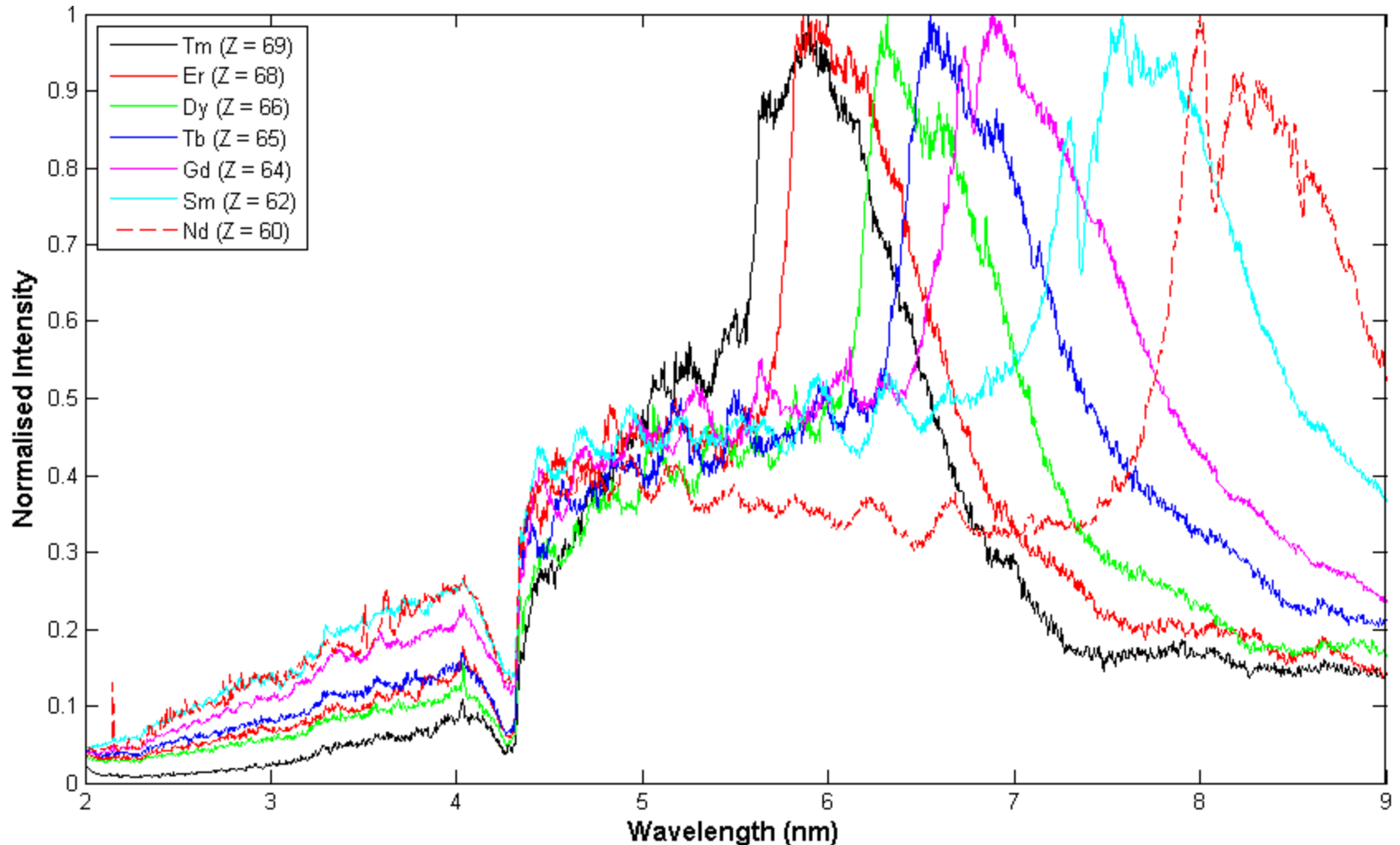
AZ

Sample of Spectra Recorded



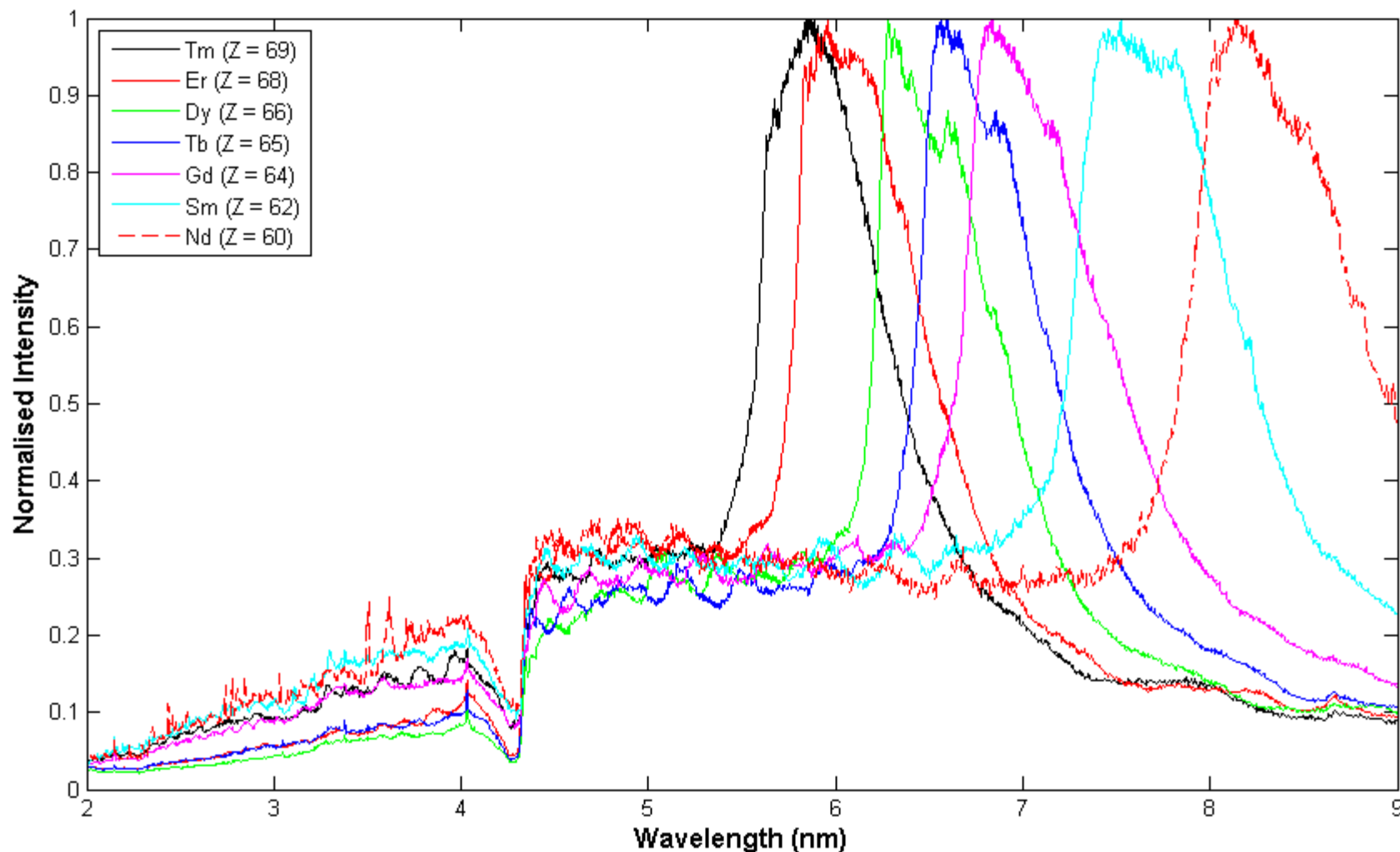
Experimentally recorded spectra of medium to high-Z elements with 8 ns, 1064 nm pulse (power density of $\sim 10^{13} \text{ Wcm}^{-2}$).

Spectra of Lanthanides using 8 ns pulse



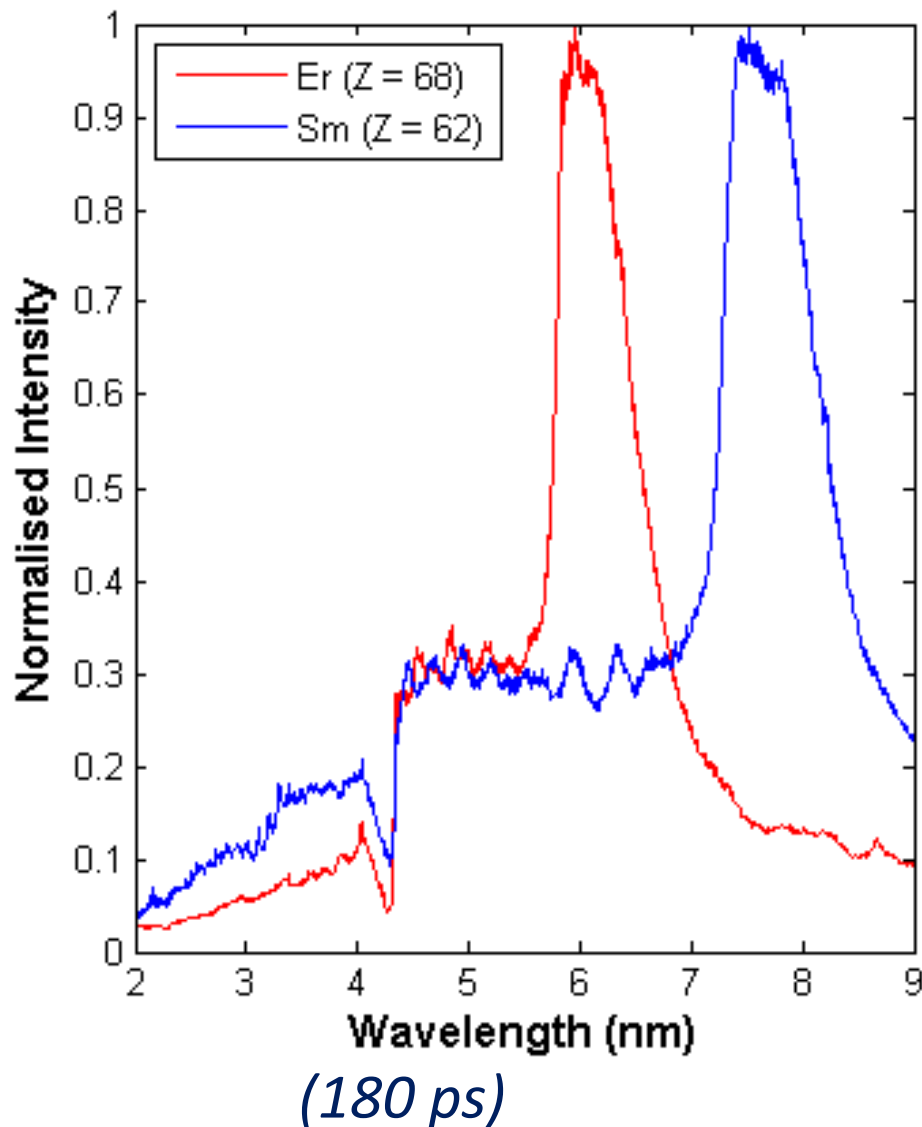
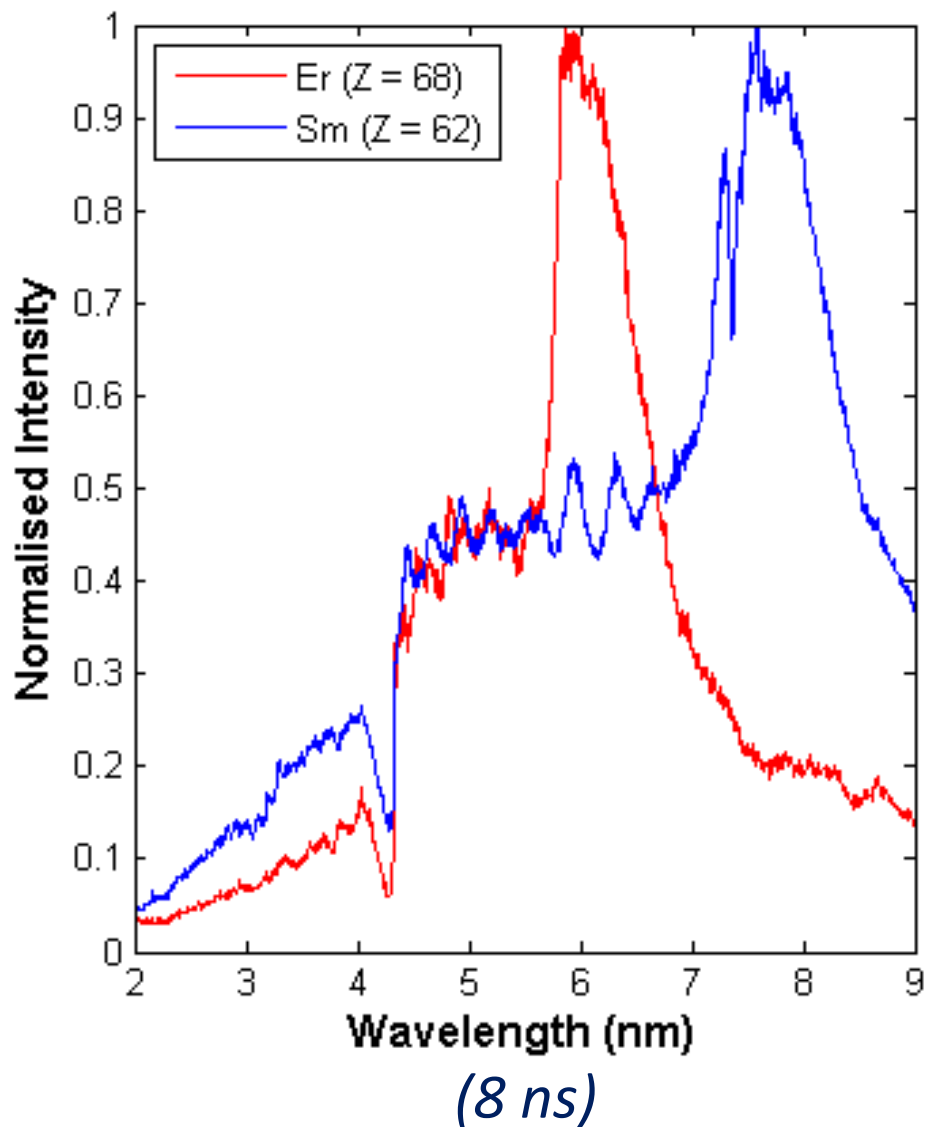
($E \sim 600 \text{ mJ}$, power density $\sim 1 \times 10^{13} \text{ Wcm}^{-2}$)

Spectra of Lanthanides using 180 ps pulse

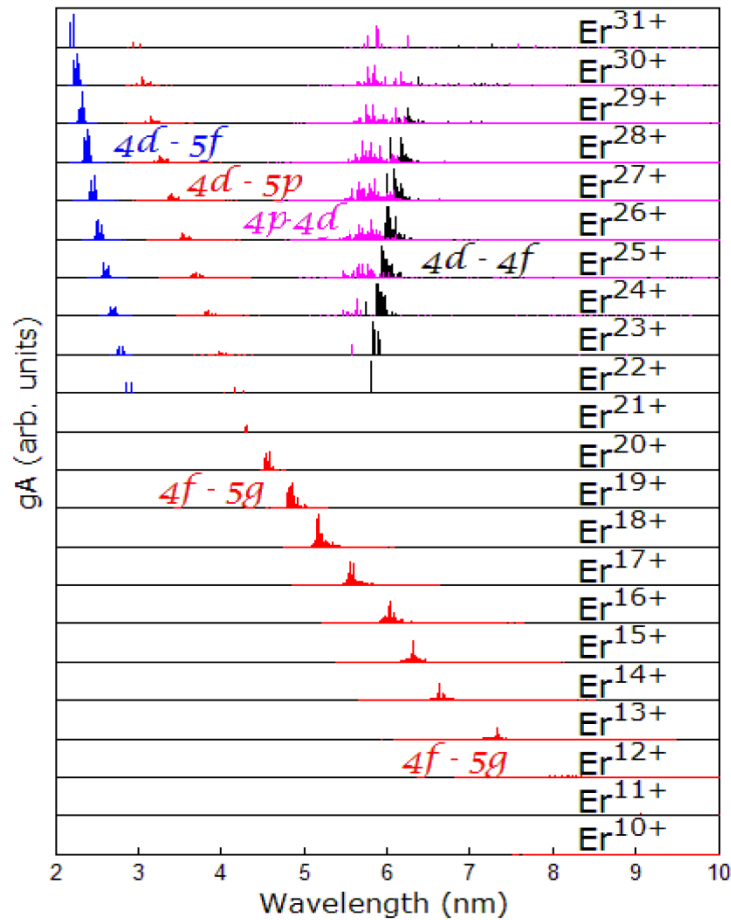


($E \sim 250$ mJ, Power Density $\sim 4 \times 10^{14}$ Wcm $^{-2}$)

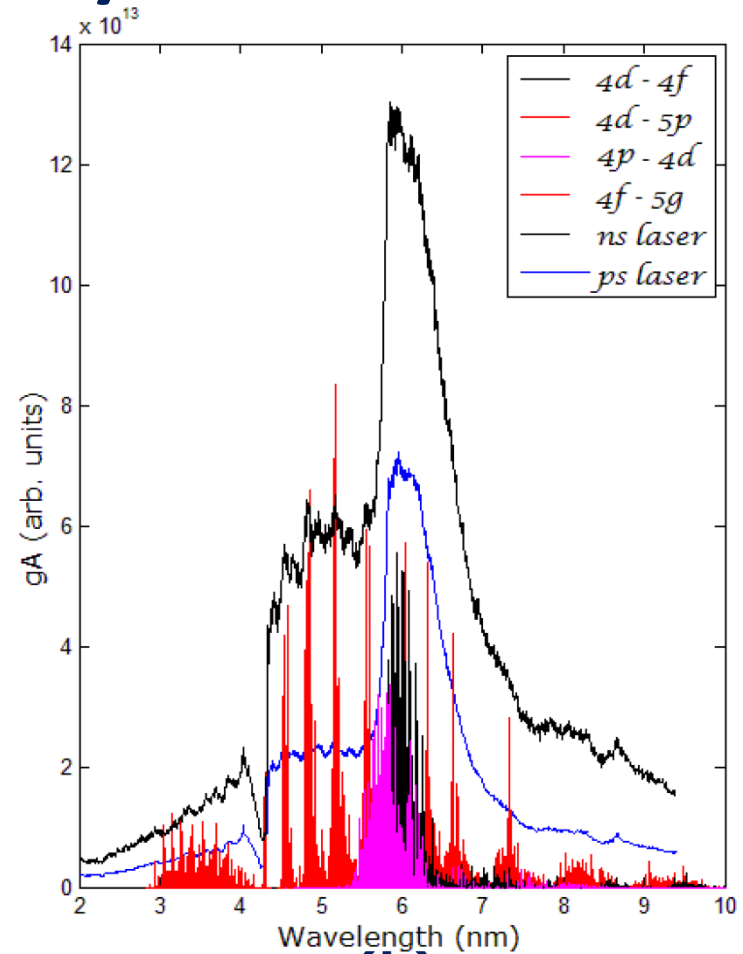
Spectra of Lanthanides (long and short pulse)



Calculated & Experimental Spectra of Er (Z=68)



(a)



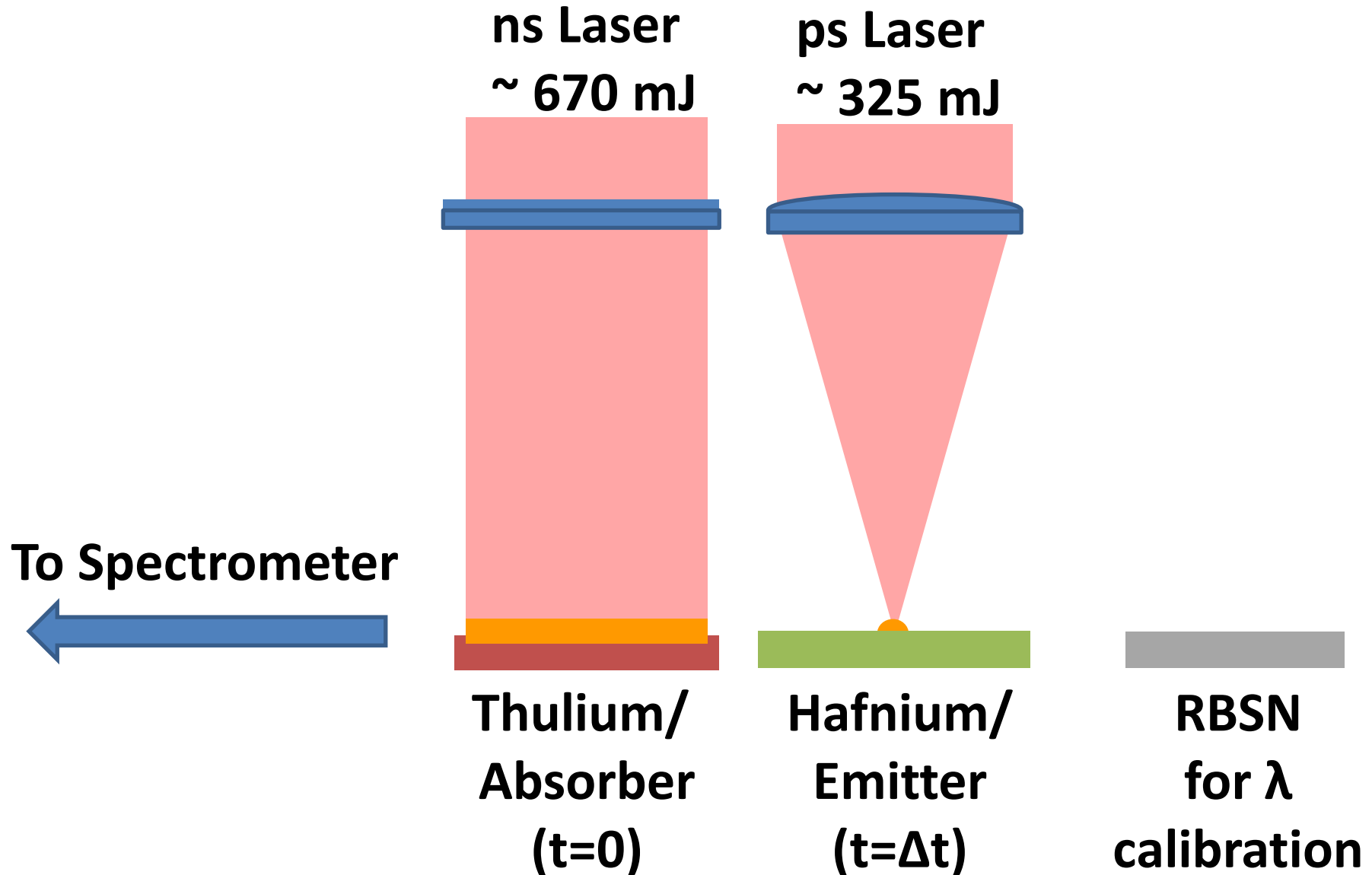
(b)

(a) shows Hartree-Fock with CI calculations performed using the Cowan Suite of Codes for erbium. Figure 8 (b) shows both the calculated and experimental spectra for an erbium target.

Outline

- Emission Spectroscopy
 - Experimental Set-up
 - Targets used
 - Spectra obtained
 - Cowan Code Calculations
- Absorption Spectroscopy
- Plasma Imaging

Photoabsorption of Thulium



Photoabsorption of Thulium

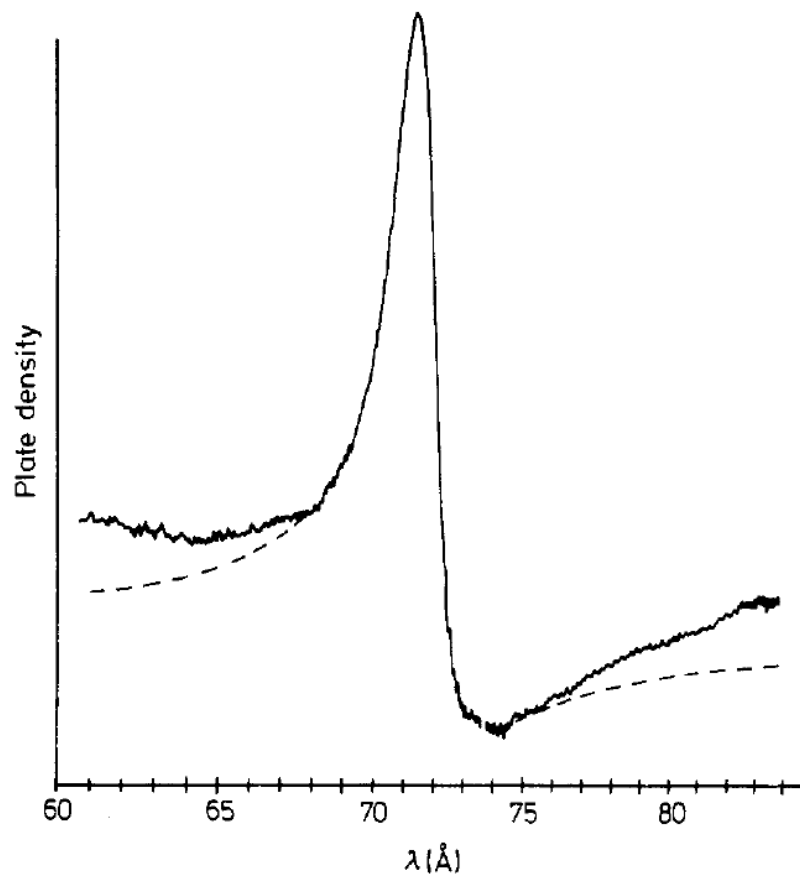
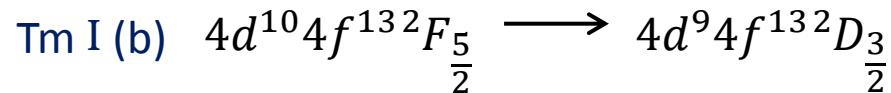
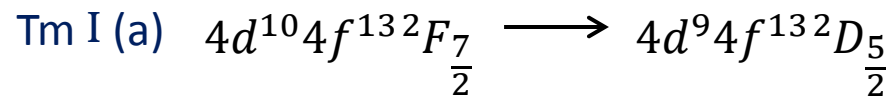
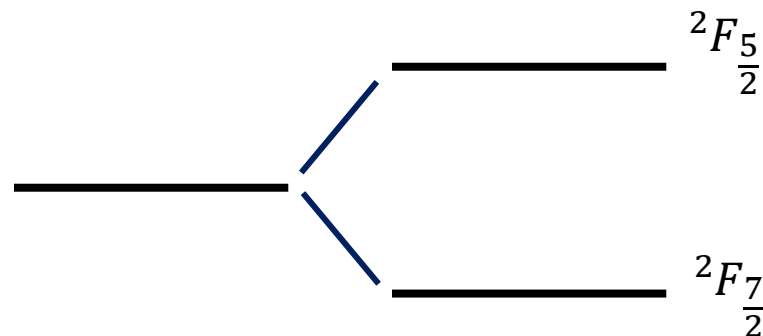


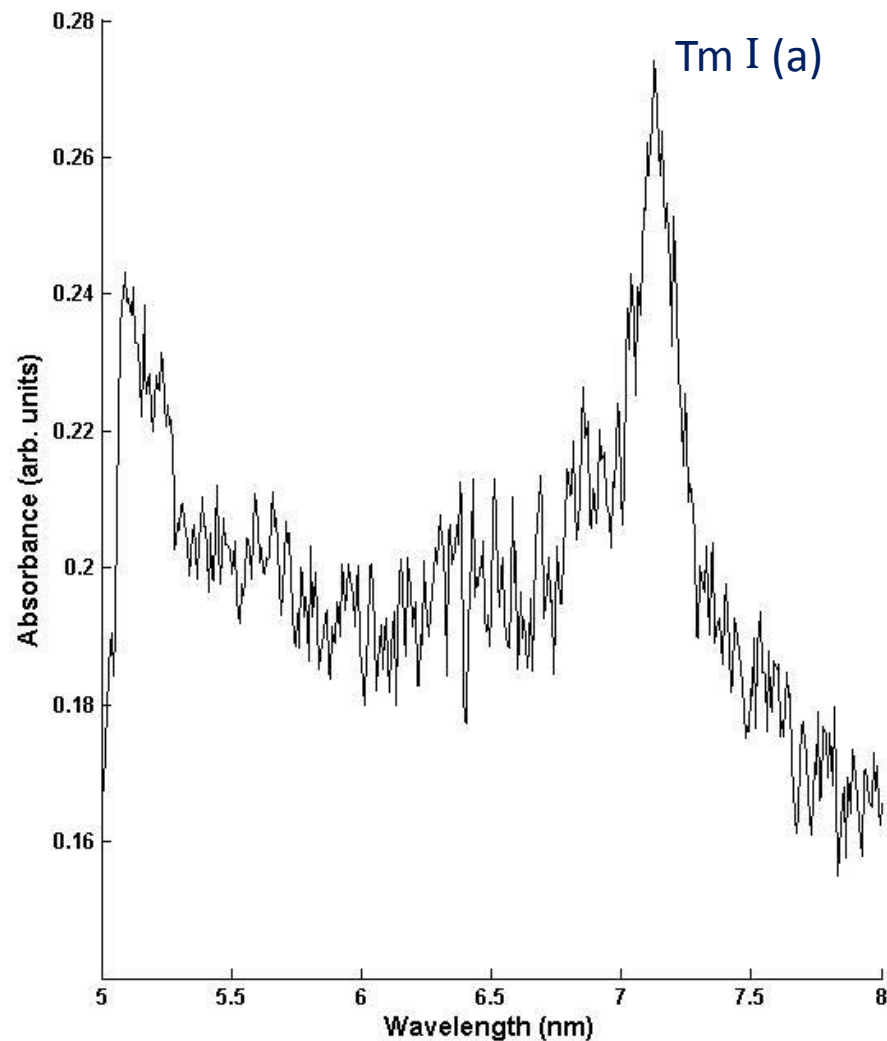
Figure 2. The 4d photoabsorption spectrum of Tm I. Full curve, data for Tm vapour; broken curve, Fano profile fit, where it deviates from the observed spectrum. Profile parameters: $q = 3.6 \pm 0.4$, $\Gamma = 2.8 \pm 0.2$ eV, $E_0 = 173.15 \pm 0.1$ eV, $\sigma_b = \text{constant } 1$, $\sigma_a(\epsilon) = \text{constant } 2$.



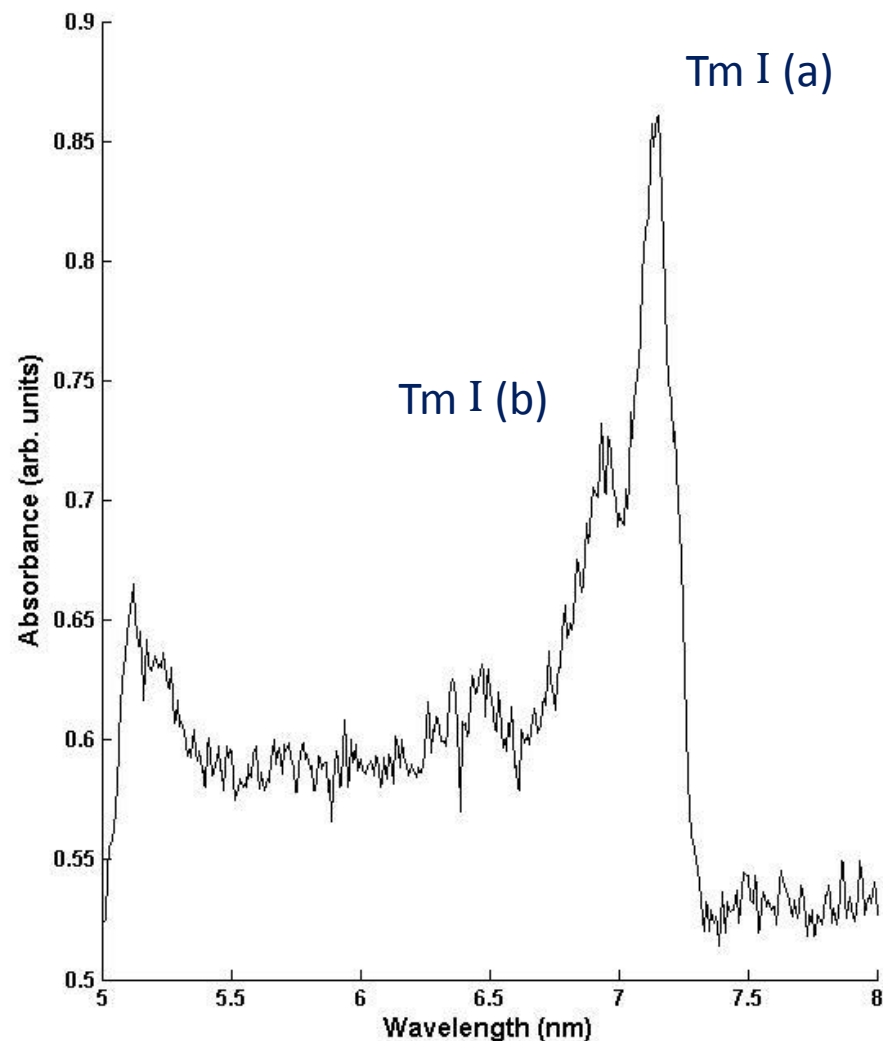
Tm I	$4p^6 4d^{10} 5s^2 5p^6 6s^2 4f^{13}$
Tm II	$4p^6 4d^{10} 5s^2 5p^6 6s 4f^{13}$
Tm III	$4p^6 4d^{10} 5s^2 5p^6 4f^{13}$
Tm IV	$4p^6 4d^{10} 5s^2 5p^6 4f^{12}$

Radtke's Paper – "On the character of the intense 4d→f resonances in atomic La and Tm"

Thulium; 4d-4f, 1000 & 400 ns

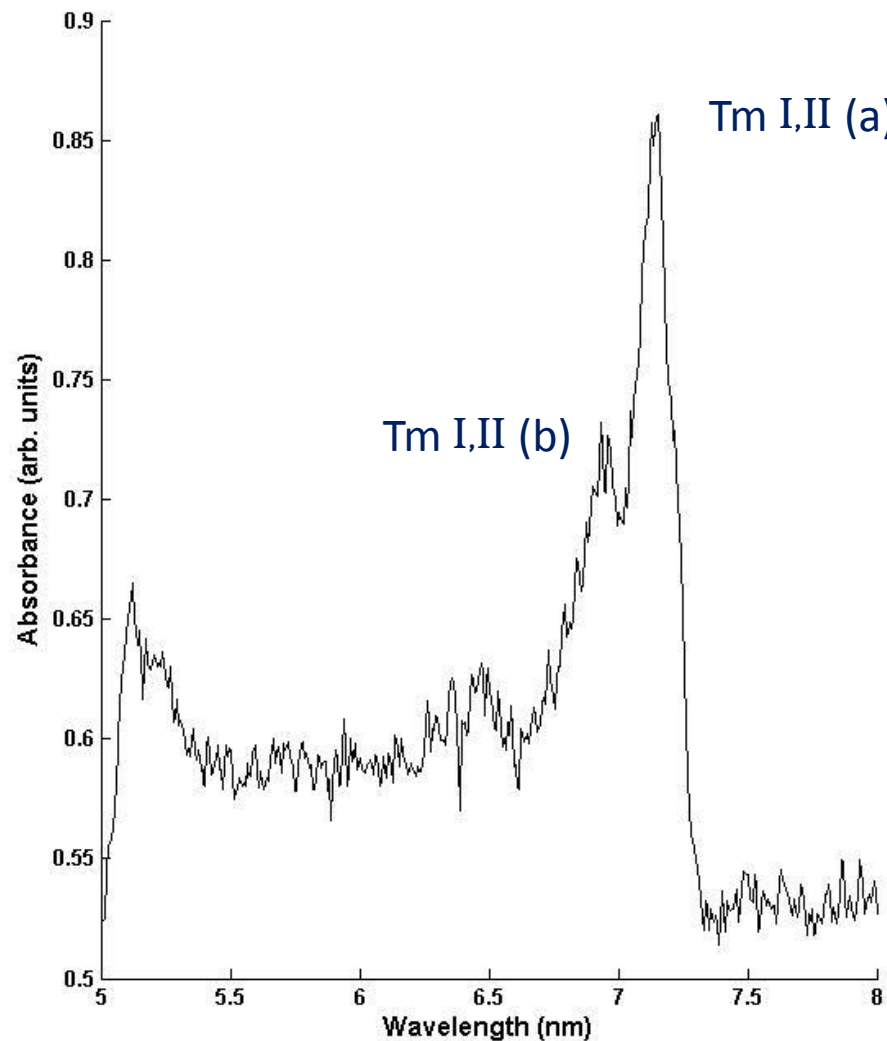


1 μs delay

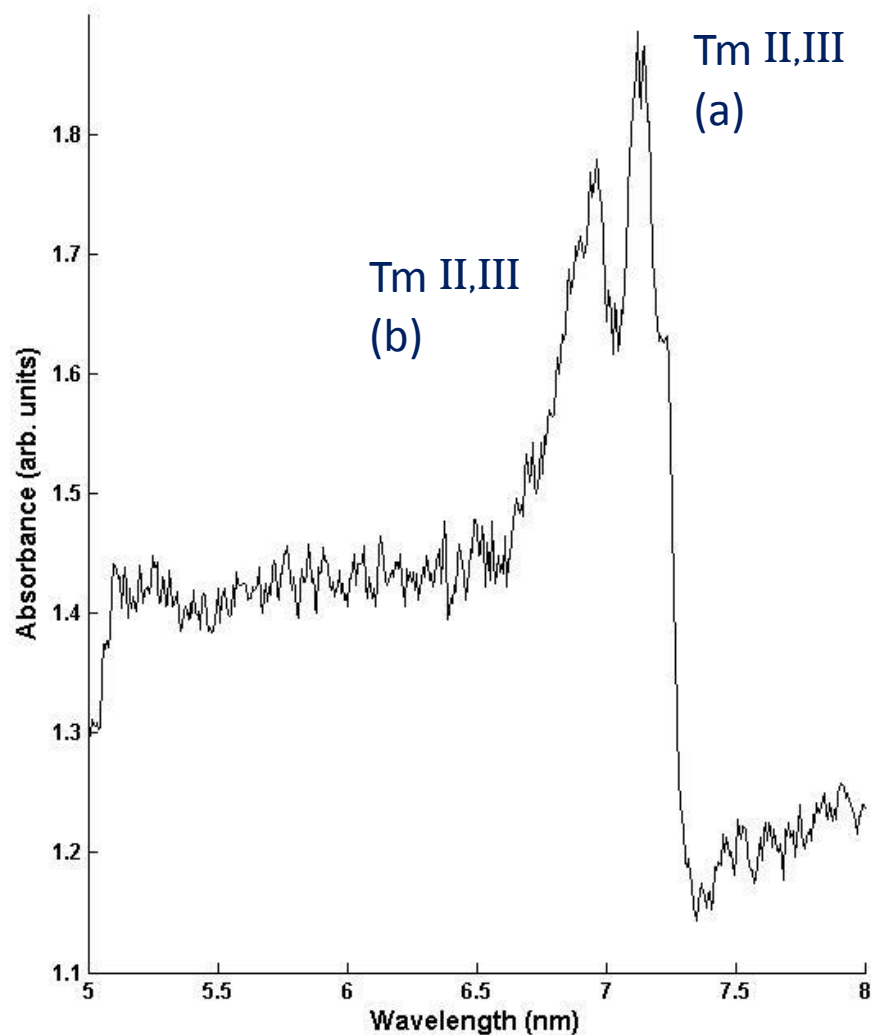


400 ns delay

Thulium; 4d-4f, 400 & 200 ns

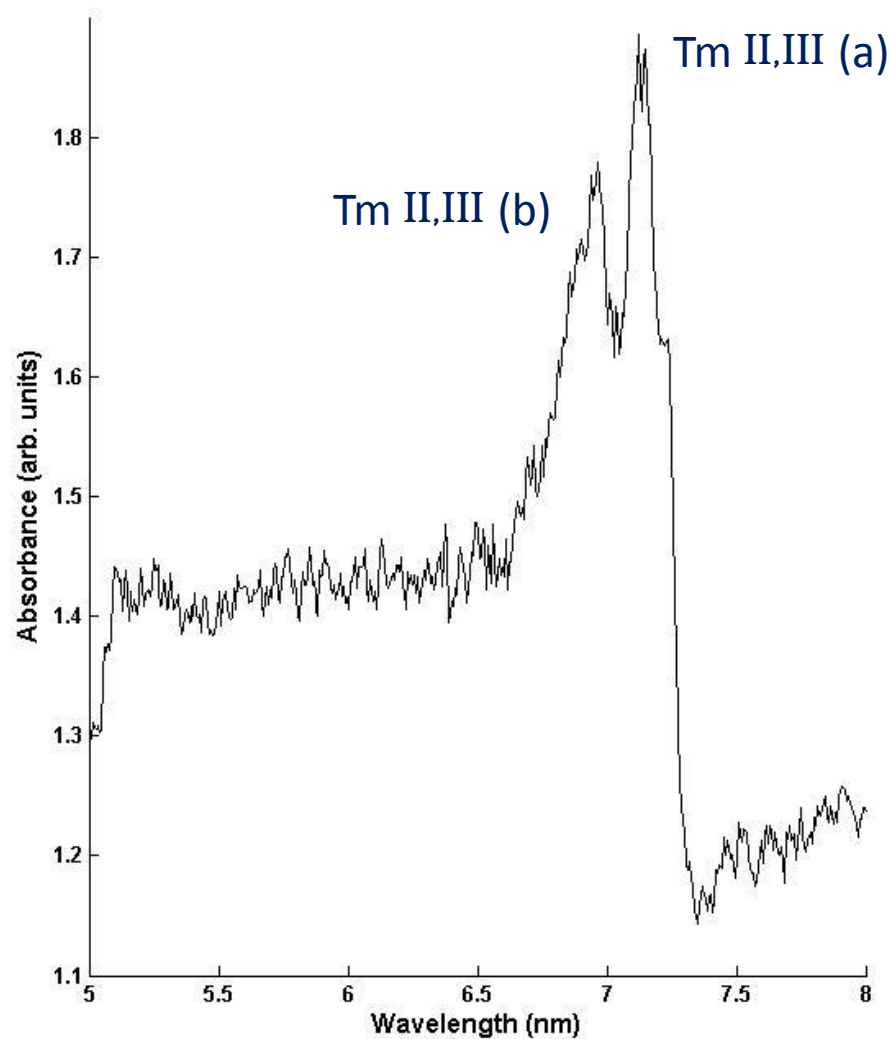


400 ns delay

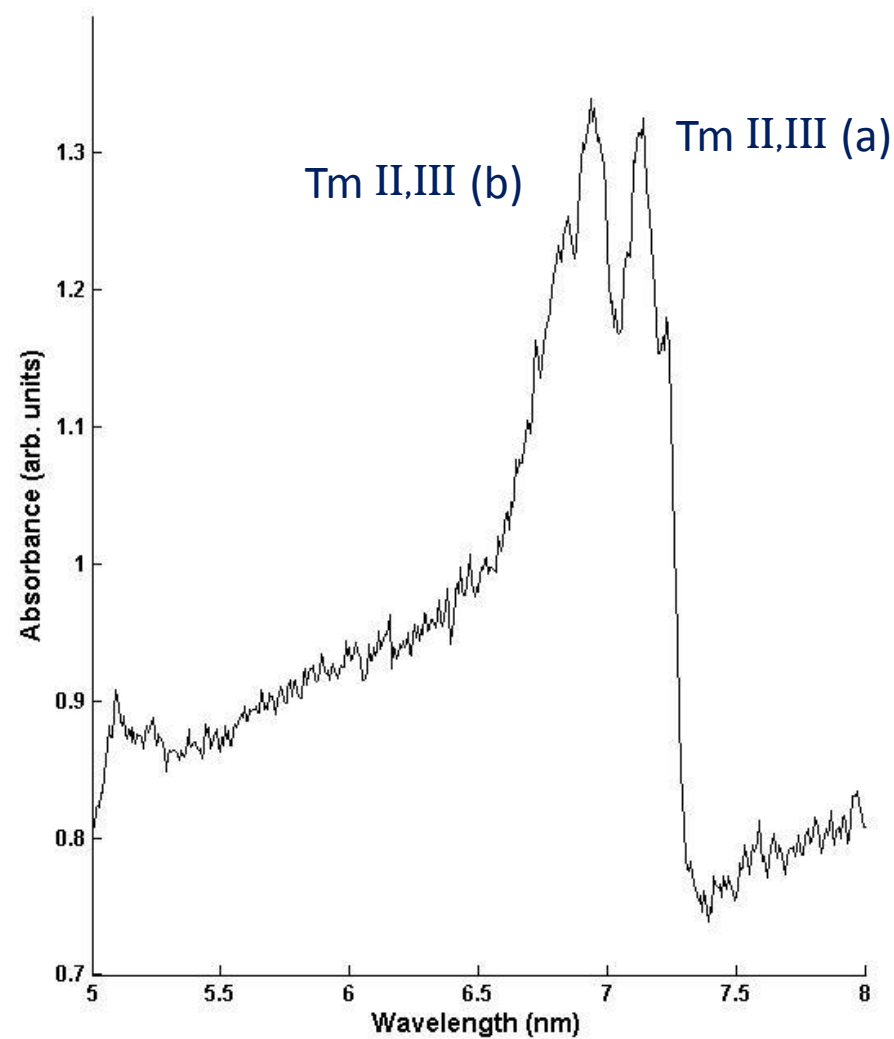


200 ns delay

Thulium; 4d-4f, 200 & 100 ns

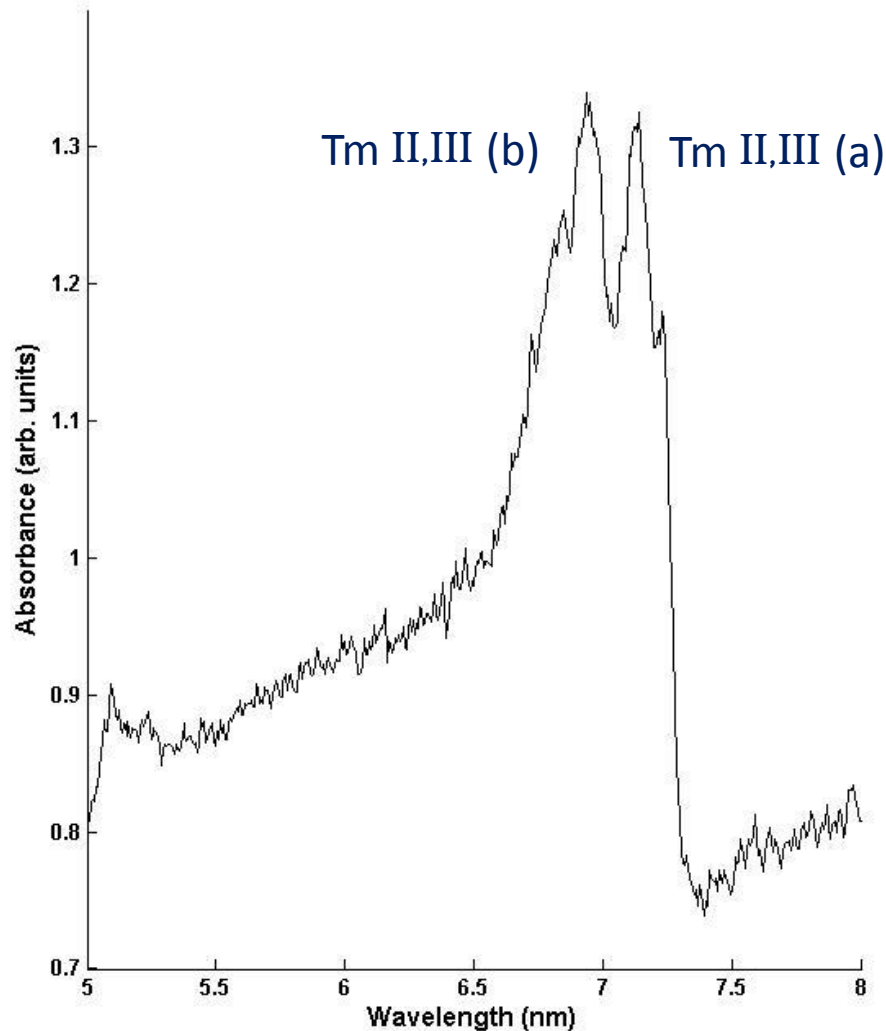


200 ns delay

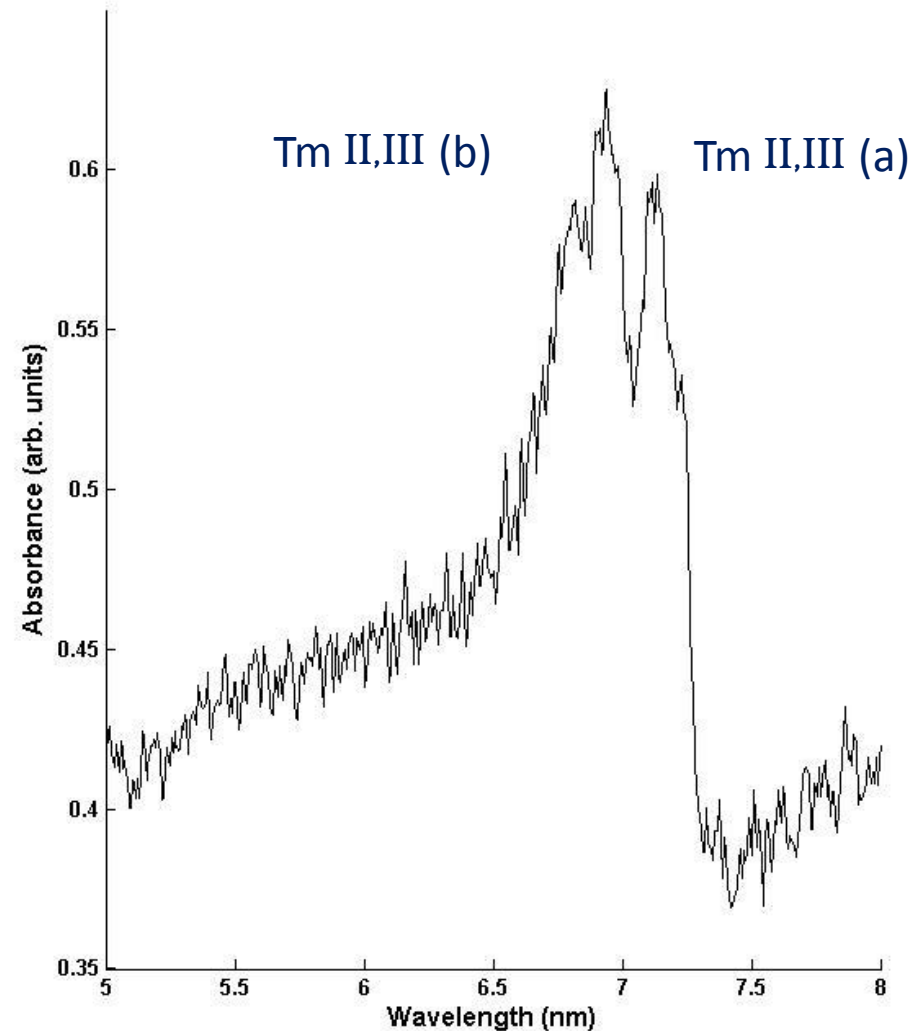


100 ns delay

Thulium; 4d-4f, 100 & 50 ns

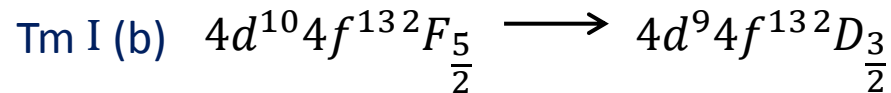
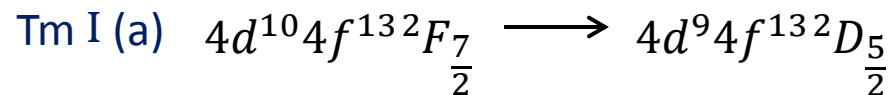
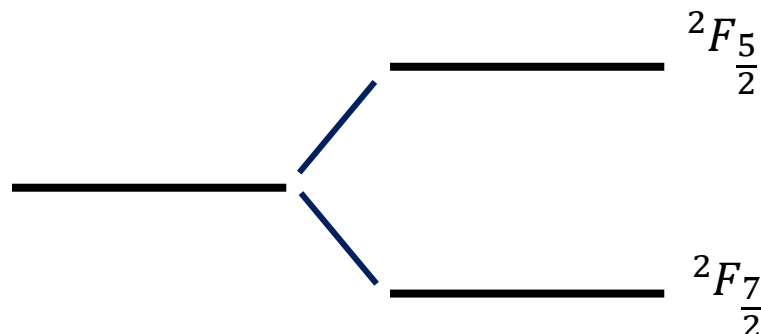
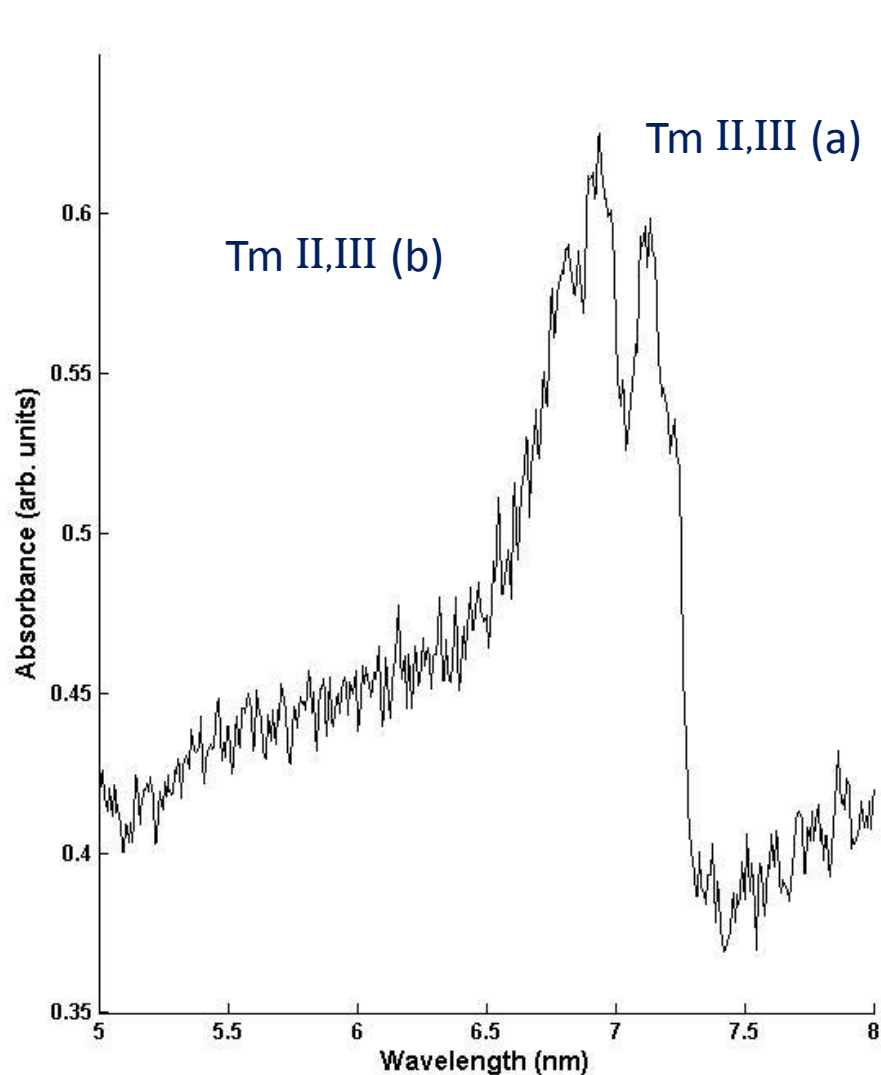


100 ns delay



50 ns delay

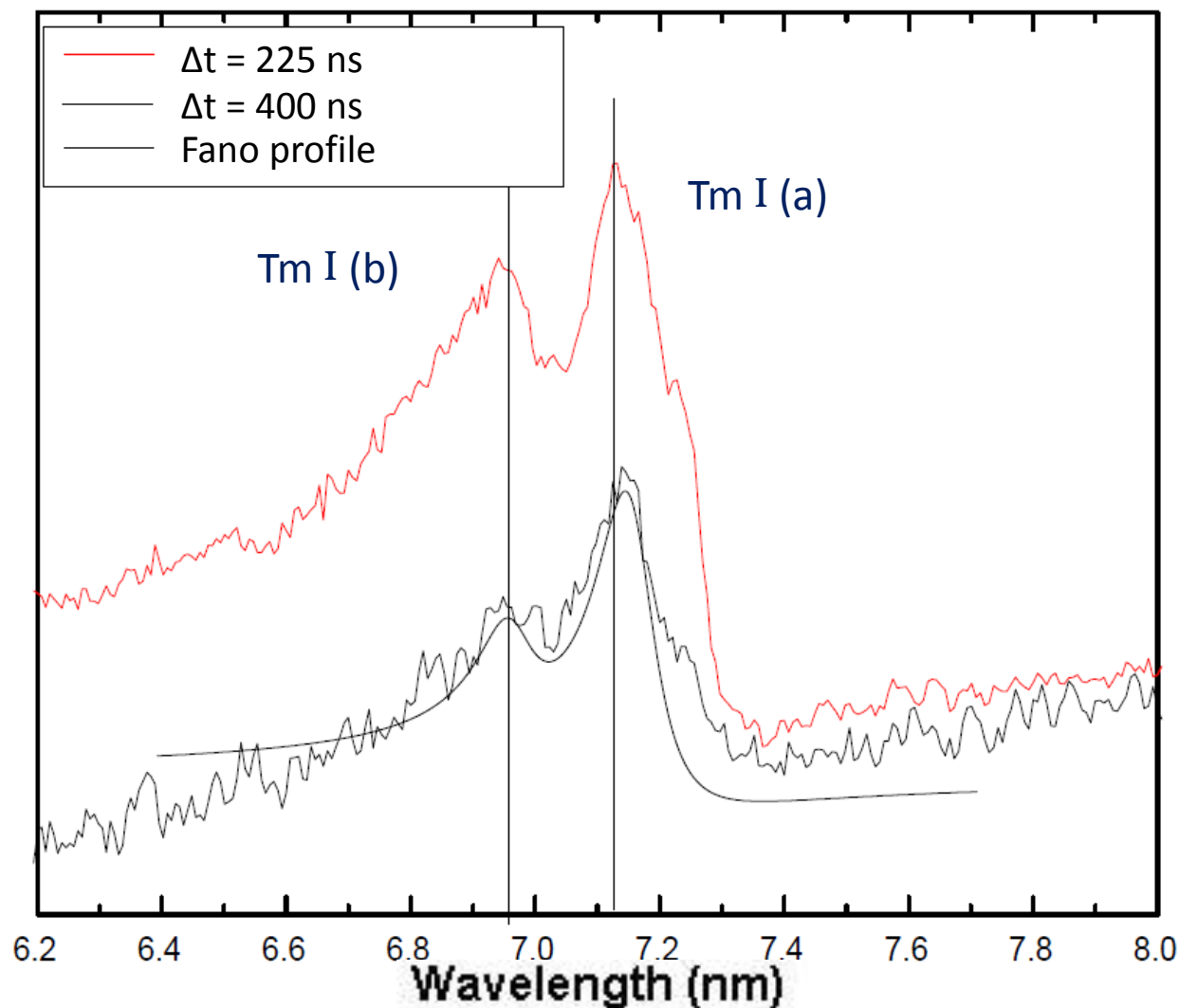
Photoabsorption of Thulium; 4d-4f transitions



Tm I	$4p^64d^{10}5s^25p^66s^24f^{13}$
Tm II	$4p^64d^{10}5s^25p^66s4f^{13}$
Tm III	$4p^64d^{10}5s^25p^64f^{13}$
Tm IV	$4p^64d^{10}5s^25p^64f^{12}$

50 ns delay

Photoabsorption of Thulium; 4d-4f transitions



Targets to be used for Absorption Spectroscopy Experiments

Spectroscopy Experiments																		
hydrogen 1 H 1.0079											helium 2 He 4.0026							
lithium 3 Li 6.941	beryllium 4 Be 9.0122											boron 5 B 10.811	carbon 6 C 12.011	nitrogen 7 N 14.007	oxygen 8 O 15.999	fluorine 9 F 18.998	neon 10 Ne 20.180	
sodium 11 Na 22.990	magnesium 12 Mg 24.305											aluminium 13 Al 26.982	silicon 14 Si 28.086	phosphorus 15 P 30.974	sulfur 16 S 32.065	chlorine 17 Cl 35.453	argon 18 Ar 39.948	
potassium 19 K 39.098	calcium 20 Ca 40.078	scandium 21 Sc 44.956	titanium 22 Ti 47.867	vanadium 23 V 50.942	chromium 24 Cr 51.996	manganese 25 Mn 54.938	iron 26 Fe 55.845	cobalt 27 Co 58.933	nickel 28 Ni 58.693	copper 29 Cu 63.546	zinc 30 Zn 65.39	gallium 31 Ga 69.723	germanium 32 Ge 72.61	arsenic 33 As 74.922	selenium 34 Se 78.96	bromine 35 Br 79.904	krypton 36 Kr 83.80	
rubidium 37 Rb 85.468	strontium 38 Sr 87.62	yttrium 39 Y 88.906	zirconium 40 Zr 91.224	niobium 41 Nb 92.906	molybdenum 42 Mo 95.94	technetium 43 Tc [98]	ruthenium 44 Ru 101.07	rhodium 45 Rh 102.91	palladium 46 Pd 106.42	silver 47 Ag 107.87	cadmium 48 Cd 112.41	indium 49 In 114.82	tin 50 Sn 118.71	antimony 51 Sb 121.76	tellurium 52 Te 127.60	iodine 53 I 126.90	xenon 54 Xe 131.29	
caesium 55 Cs 132.91	barium 56 Ba 137.33	57-70 *	lutetium 71 Lu 174.97	hafnium 72 Hf 178.49	tantalum 73 Ta 180.95	tungsten 74 W 183.84	rhenium 75 Re 186.21	osmium 76 Os 190.23	iridium 77 Ir 192.22	platinum 78 Pt 195.08	gold 79 Au 196.97	mercury 80 Hg 200.59	thallium 81 Tl 204.38	lead 82 Pb 207.2	bismuth 83 Bi 208.98	polonium 84 Po [209]	astatine 85 At [210]	radon 86 Rn [222]
francium 87 Fr [223]	radium 88 Ra [226]	89-102 * *	lawrencium 103 Lr [262]	rutherfordium 104 Rf [261]	dubnium 105 Db [262]	seaborgium 106 Sg [266]	bohrium 107 Bh [264]	hassium 108 Hs [269]	meitnerium 109 Mt [268]	ununnium 110 Uun [271]	ununium 111 Uuu [272]	unubium 112 Uub [277]						
													ununquadium 114 Uuq [289]					

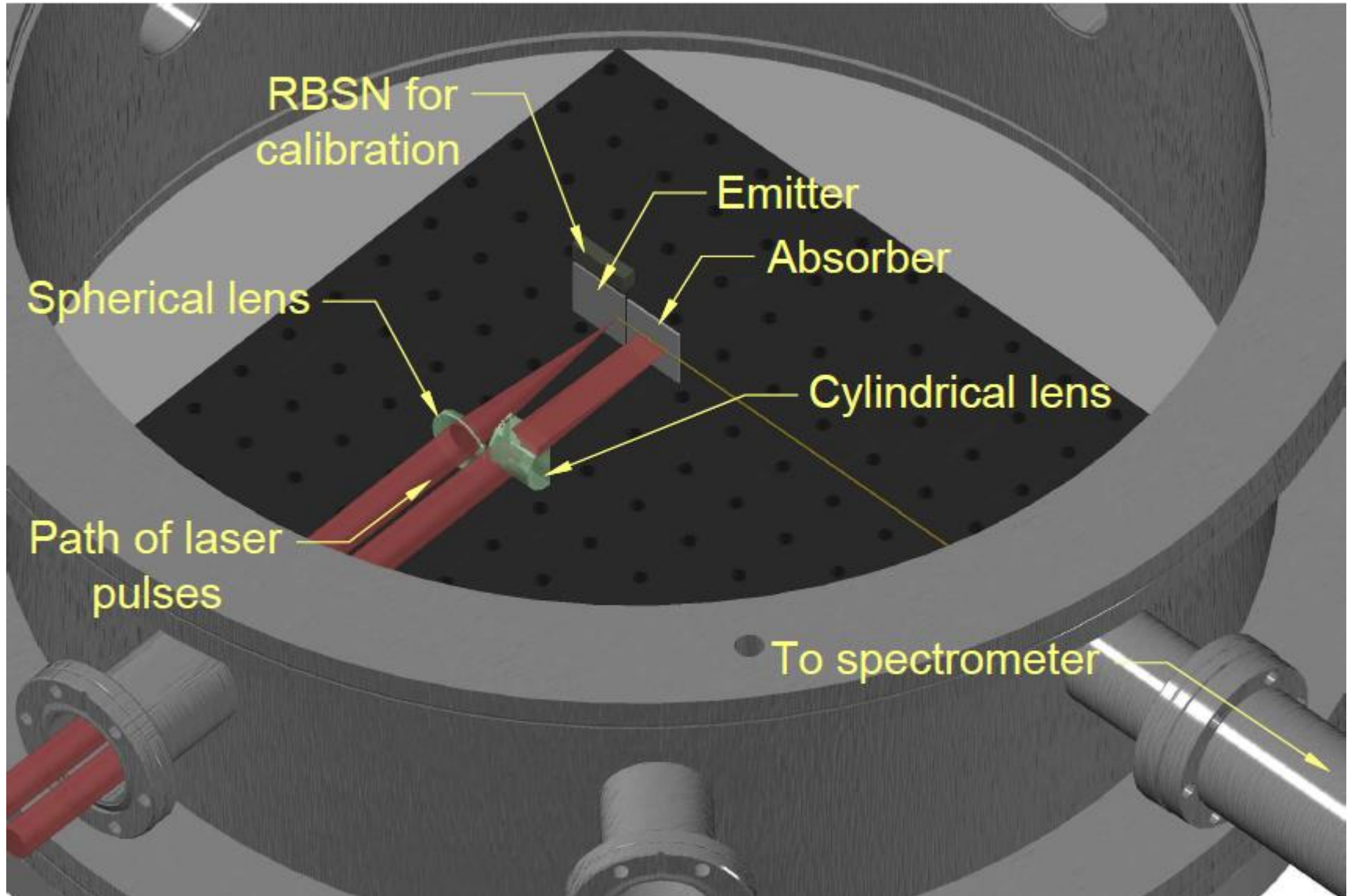
* Lanthanide series

* * Actinide series

lanthanum 57 La 138.91	cerium 58 Ce 140.12	praseodymium 59 Pr 140.91	neodymium 60 Nd 144.24	promethium 61 Pm [145]	samarium 62 Sm 150.36	europium 63 Eu 151.96	gadolinium 64 Gd 157.25	terbium 65 Tb 158.93	dysprosium 66 Dy 162.50	holmium 67 Ho 164.93	erbium 68 Er 167.26	thulium 69 Tm 168.93	ytterbium 70 Yb 173.04
actinium 89 Ac [227]	thorium 90 Th 232.04	protactinium 91 Pa 231.04	uranium 92 U 238.03	neptunium 93 Np [237]	plutonium 94 Pu [244]	americium 95 Am [243]	curium 96 Cm [247]	berkelium 97 Bk [247]	californium 98 Cf [251]	einsteinium 99 Es [252]	fermium 100 Fm [257]	mendelevium 101 Md [258]	nobelium 102 No [259]

AZ

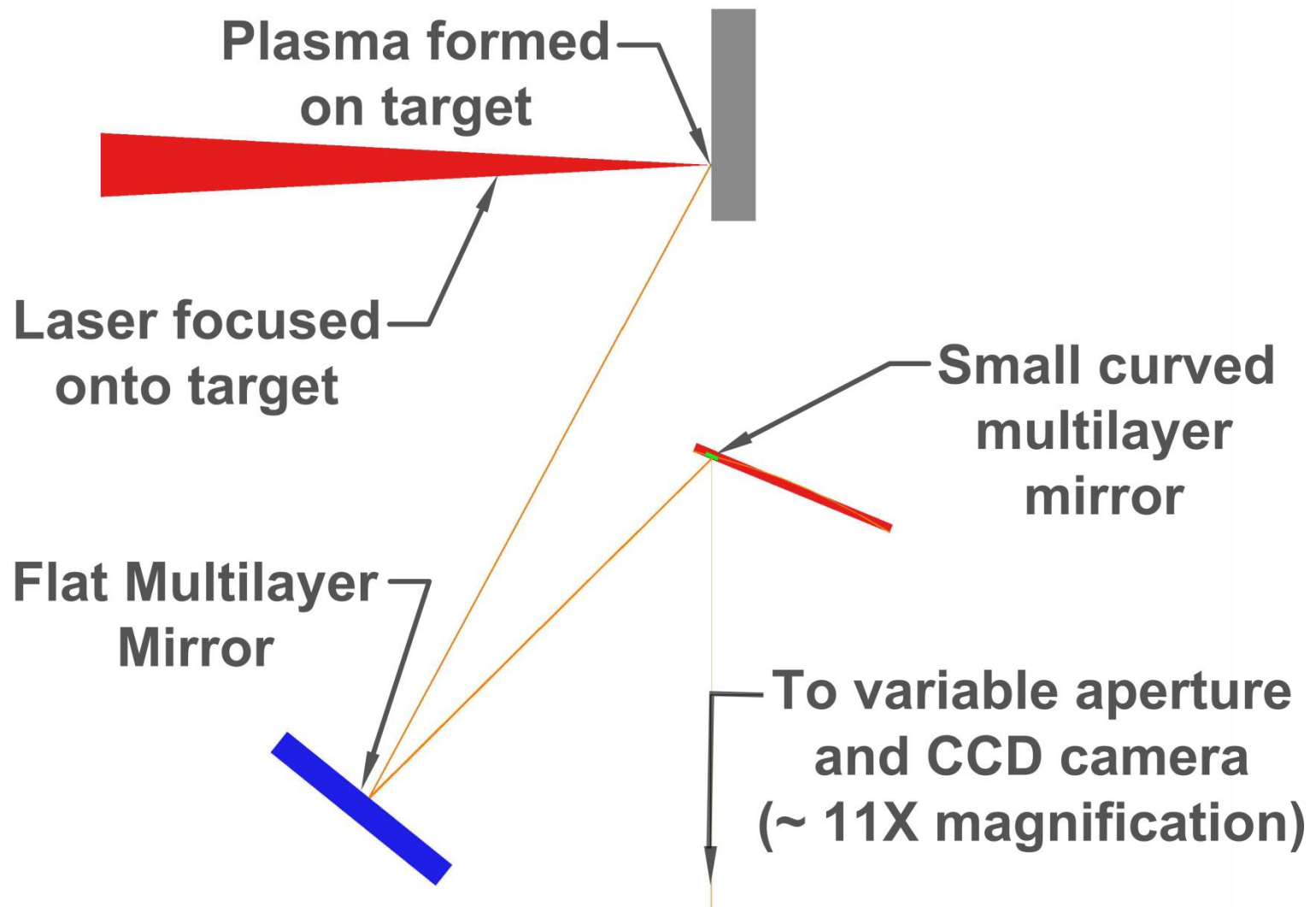
Photoabsorption Set-up



Outline

- Emission Spectroscopy
 - Experimental Set-up
 - Targets used
 - Spectra obtained
- Absorption Spectroscopy
- Plasma Imaging

Imaging Plasmas in the Soft X-Ray Region Using Multilayer Optics



Future Work & Conclusions

Main focus will be on producing *small-diameter, high-brightness* plasma sources in the soft X-ray region by;

- Various high-energy, pulsed lasers with pulse lengths from 30 fs to 100 ns will be used to create plasmas for comparison to the 180-ps laser.
- Imaging Plasmas in the soft X-ray region with pinhole camera and multilayer mirrors.
- Absorption Spectroscopy of low-Z elements.

Thank You!

This project is funded by the Irish Research Council and Intel Ireland under the Enterprise Partnership Scheme. I would like to thank the technical officers in the mechanical and electronic workshops for their support.

